

CHEMISTRY

A EUROPEAN JOURNAL

15/2

2009

Metal Oxide Nanoparticles

Legend:

- Co²⁺ + Fe³⁺
- Ce⁴⁺
- Ti⁴⁺
- Al³⁺
- Zn²⁺
- Cu²⁺
- Na⁺

Labels in Nanoparticle Insets: CuO, ZnO, Al₂O₃, CoFe₂O₄, CeO₂, TiO₂

Labels in Diagram: alkali metal oxide solution, alkali metal oxides

Now
Weekly!

A Journal of



Supported by
ACES

Concept

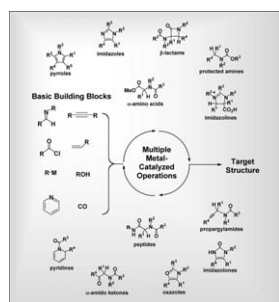
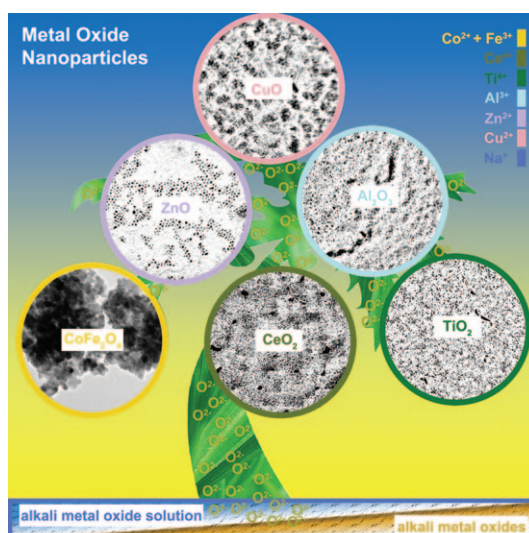
Metal-Catalyzed One-Step Synthesis: Towards Direct Alternatives to Multistep Heterocycle and Amino Acid Derivative Formation

B. A. Arndtsen

WILEY-VCH

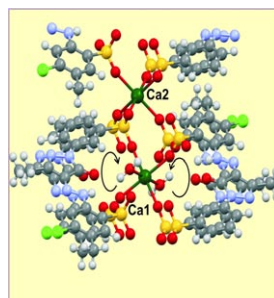
Metathesis of metal oxides...

... provides a novel route to produce nanoscale oxide materials in an inexpensive and highly efficient way. This has been explained pictorially by M. A. Morris et al. on page 440 ff. using an oxide 'tree': Oxide anions are released from the 'pool' (alkali metal oxides serve as the oxide ion source) and are absorbed by the 'root' of the 'tree' and then transferred via the 'trunk' to form the 'fruits' (metal oxide nanoparticles) at the top of the 'tree'. The different colors refer to the different condensation states of each oxide.



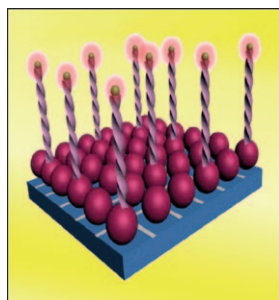
Multicomponent Reactions

In the Concept article on page 302 ff., B. A. Arndtsen demonstrates how the reactivity of transition-metal catalysis can be exploited to build up families of useful products in a single step from available building blocks. These reactions rely upon an array of metal-based reactions to occur in a sequential fashion in order to selectively bring these units together. By gaining an understanding of each of these steps, and the various side reactions within these transformations, selective metal-catalyzed routes to a broad range of products can be developed.



X-ray Diffraction

In their Communication on page 338 ff., M. U. Schmidt et al. show that crystal structures, even of molecular salts with unknown water content, can be solved from a laboratory X-ray powder data measured in a routine manner.



Gold Nanoparticles

The frontispiece picture shows an artist's view of the nanoparticle surface-energy transfer (NSET) probe for sensing hepatitis C virus RNAs, which can serve as a nanoscopic ruler for improving crucial analytical criteria, such as detection limits, sensitivity, selectivity, and dynamic range relative to commercial systems; see the Full Paper by P. C. Ray et al. on page 342 ff. for more details.

Supported by
ACES

Chemistry—A European Journal is jointly owned by the 14 Chemical Societies shown above and published by Wiley-VCH. This group of Societies has banded together as Chemistry Publishing Society (ChemPubSoc) Europe for its combined publishing activities. The journal is also supported by the Asian Chemical Editorial Society (ACES).